

PII: S0040-4020(97)00037-9

1,3-Dipolar Cycloaddition Approach to Indolic Aza-analogues of Cephalotaxine

Miklós Nyerges, Mónika Rudas, István Bitter and László Tőke*

Research Group of the Hungarian Academy of Sciences, Department of Organic Chemical Technology, Technical University of Budapest, H-1521 Budapest P.O.B. 91, Hungary

Csaba Szántay, Jr.,*

Chemical Works of Gedeon Richter Ltd., Spectroscopic Research Division, H-1475 Budapest 10, P.O.B 27, Hungary

Abstract: An indolic aza-analogue (14) of cephalotaxine has been prepared stereoselectively using 1,3-dipolar cycloaddition of azomethine ylide as a key step. The Pictet-Spengler reaction of the amine 10 resulted in the formation of an unusual heterocyclic product (11). The structure and stereochemistry of 11 and 14 were studied in detail by n.m.r. spectroscopic methods.

© 1997 Elsevier Science Ltd. All rights reserved.

INTRODUCTION

Cephalotaxine 1, the parent alkaloid produced by conifers of the Cephalotaxus genus has a unique skeleton with an unusual 1-azaspiro[4.4]nonane moiety fused to a benzazepine system. Synthetic studies on Cephalotaxus alkaloids have attracted the attention of several research groups interested in the antileukaemic properties of naturally occurring ester derivatives of cephalotaxine 1, eg. harringtonine 2. Interest is now turning towards modified ring systems. For example Laronze et al. reported the preparation of the indolic analogue 3. In addition, the methylenedioxy- and dimethoxy-derivatives of the hexahydrocyclopenta[e]phenanthridine ring system as well as the more highly substituted analogues, notably hydroxy-and oxo-derivatives, have been identified as important targets. We have been exploring an entirely new approach towards the aza-analogues of cephalotaxine 1 with modified pentacyclic skeletons. Here we describe our work on the synthesis of the new aza-indolic cephalotaxine skeleton 14 which was obtained stereospecifically in an eight-step sequence starting from the indolo nitro-ethylene 4 as outlined in Schemes 2 and 4.

Scheme 1

RESULTS AND DISCUSSIONS

The azomethine ylide prepared from imine 5 by N-metallation⁹ gave a 3:1 mixture of diastereometric pyrrolidine derivatives from which the major isomer 6 was obtained in pure form by simple crystallization in 51 % yield.⁸ Acetylation of 6 gave 7 whose Michael reaction with methyl acrylate afforded the adduct 8 as a single isomer. The reductive cyclization followed by LiAlH₄ reduction of the nitro-ester 8 gave rise exclusively the key compound 10 with the correct stereochemistry at the spiro-centre.

Scheme 2 (a) CH₃CN, AgOAc, Et₃N, r.t.; (b) Ac₂O, Py; (c) CH₂=CHCO₂Me, Triton B; (d) Zn, HCl, EtOH; (e) LiAlH₄, THF;

The amine 10 was allowed to react with formaldehyde under Pictet-Spengler conditions in benzene using HCl as catalyst. This reaction led to an unusual six-membered heterocycle (11) instead of the expected product (14). The same result was obtained by using Eschenmoser's salt.¹⁰

Scheme 3 (a) (CH₂O), HCl, benzene, reflux or (CH₃)₂N⁺=CH₂I, THF, r.t.

This reaction path can be rationalized in view of the well-known theory of the mechanism of Pictet-Spengler reaction in the formation of β -carbolines. In our case the spiroindolenine intermediate did not rearrange to the corresponding β -carboline derivative because of the faster trapping by nucleophilic attack of the hydroxyl group at the electrophilic α -indole carbon.

To avoid this unwanted reaction the hydroxymethyl group in 10 was protected with *tert*-butyldimethylsilyl group, then cyclized to the indolo-cephalotaxine analogue 13 followed by deprotection with tetrabutylammonium fluoride to give 14 in 10.6 % overall yield (for 8 steps starting from 4).

ROCH₂

$$\stackrel{\text{Fi}}{\longrightarrow}$$
 $\stackrel{\text{Ph}}{\longrightarrow}$
 $\stackrel{\text{CH}_3}{\longrightarrow}$
 $\stackrel{\text{CH}_3}{\longrightarrow}$
 $\stackrel{\text{13 R = TBDMS}}{\longrightarrow}$
 $\stackrel{\text{12 R = TBDMS}}{\longrightarrow}$
 $\stackrel{\text{C}}{\longrightarrow}$
 $\stackrel{\text{ROCH}_2}{\longrightarrow}$
 $\stackrel{\text{Ph}}{\longrightarrow}$
 $\stackrel{\text{C}}{\longrightarrow}$
 $\stackrel{\text{C}}{\longrightarrow}$

Scheme 4 (a) TBDMSCl, DBU, CH₃CN, (b) (CH₃)₂N⁺=CH₂I⁻, THF, r.t. (c) Bu₄NF/SiO₂, THF.

The structures of the above discussed compounds were elucidated by detailed n.m.r. studies. The depicted configurations were verified by homonuclear n.O.e. measurements as listed in the Experimental section. The n.m.r characteristics of compounds 11 and 14 will be given below.

Compound 11. The ¹H and ¹³C n.m.r. assignments of compound 11, in conjuction with its structure determination, were based on the concerted use of high-field one and two-dimensional n.m.r. methods (DQFCOSY, HSQC, HMBC, ¹H{¹H} and ¹³C{¹H} n.O.e. difference). ¹H and ¹³C n.m.r. data for compound 11 are collected in Table 1. and verify the structure of 11 in a straightforward manner.

| No. | δН | $[J_{H,H}(Hz)]$ | ¹ H{ ^t H} n.O.e. ^a | <i>8</i> € | НМВС ^ь |
|----------|---------------------|-------------------------------|--|------------|-------------------|
| 2β | 1.97 m | | | 53.3 | |
| 2α | 2.66 m | | | | |
| 3β | 1.15 m | | | 24.8 | |
| 3α | 1.76 m | | 3β , 4α , 2α , 10α | | |
| 4β | 2.02 m | | | 36.2 | |
| 4α | 2.14 m | | 4β , 6 , 10α, 5' | | |
| 5 | | | | 83.5 | 10α |
| 6 | 2.68 s ^c | | | 55.2 | |
| 7 | 2.68 s ^c | | | 55.8 | |
| 9 | 3.42 s | | 2" , 6" , 7 , 12x, 4 β , 4α , 13 | 78.5 | |
| 10α | 2.93 d | $[J_{10\beta,10\alpha}=10.5]$ | 10β , 5', 2 α , 4 α , 3 α | 67.2 | |
| 10β | 3.36 br d | | 10α, 2' , 2", 6" | | |
| 11α | 3.74 dd | $[J_{11\alpha,7}<1]$ | 11β, 7 | 58.9 | |
| 11β | 3.84 d | $[J_{11\beta,7}=1]$ | 2', 7, 12y, 13, 11α | | |
| | | $[J_{11\beta,11\alpha}=12.5]$ | | | |
| 12x | 2.52 dq | | | 41.6 | |
| 12y | 2.70 dq | | | | |
| 13 | 0.77 t | | | 8.3 | |
| NMe | 2.83 s | | 11α, 2',8' | 31.4 | _ |
| 1" | | | | 139.3 | 9, 3", 5" |
| 2", 6" | 7.49 br | | 2', 9 , 10β , $12x$, 2β , 4β , 3β , 13 | 130.0 | |
| 3", 5" | 7.24-7.30 m | | | 127.1 | |
| 4" 21 | 7.24-7.30 m | | | 127.1 | |
| 2' 3' | 4.93 s | | 2° , 6° , 11β , 10β , NMe | 97.2 | 51. 10 |
| | | | | 50.9 | 5', 10α |
| 4' | 7.02.11 | | | 135.9 | 6', 8', 10α |
| 5' | 7.03 dd | | 10α , 6 , 4α , $\underline{4\beta}$, 3α , 6' | 122.3 | |
| 6' 7' | 6.68 td | | | 118.0 | |
| 8' | 7.09 td | | SI NIM. | 127.8 | |
| 8' 9' | 6.46 dd | | 7', NMe | 106.0 | 51 71 NIMA |
| 7 | | | | 150.0 | 5', 7', NMe |

^a Different characters are used as follows. Small: weak enhancement (n.O.e.<2%); normal: medium enhancement (2%<n.O.e.<6%); bold: strong enhancement (n.O.e. >6%), underlined: small negative n.O.e. (three-spin effect). ^b The indicated protons are those having $^{2.3}J_{\rm C.H}$ correlations with the relevant quaternary carbon as detected in an HMBC experiment. ^c The H-6 and H-7 resonances coincide and give a non-first order singlet.

Table 1 ¹H and ¹³C n.m.r. chemical shifts, selected H-H couplings and measured n.O.e. and HMBC connectivities for compound 11

The measured homonuclear n.O.e. connectivities are in full agreement with the geometry of the molecule as illustrated schematically in Fig. 1 (all mobile rings as well as the NEt group are denoted according to their dominant conformation). The n.O.e.s between H-7, H-9 and H-6 (Table 1) confirm the configuration of the relevant carbons as shown in Fig. 1. It is noted that the geminal coupling $J_{10\beta,10\alpha}$ has characteristically small absolute value (10.5 Hz) due to the antiperiplanar effect of the N(1) lone pair on H-10 α . Some of the

¹H resonances (most notably H-2",6" and H-10β) are exchange-broadened (at 500 MHz) due to the hindered rotation of the C(9)-phenyl ring.

| No. | δН | $[J_{H,H}(Hz)]$ | ¹ H{ ¹ H} n.O.e. ^a | 8 C | НМВС⁰ |
|--------------|----------|---|--|------------|---------------------------------|
| 2β | 2.49 ddd | | 3β, 2 α, 2", 6" | 55.9 | |
| 2α | 2.15 ddd | | | | |
| 3β | 1.04 m | | | 23.0 | |
| 3α | 1.28 m | | | | |
| 4β | 2.13 dd | | | 38.9 | |
| 4α | 1.78 ddd | | 4β, 6, 3α, 5' | | |
| 5 | | | | 71.9 | 6, 9, 10α |
| 6 | 3.74 d | $[J_{6,7}=9.5]$ | 3α, 4α, 2α, 7, 11x, 9, 5' | 43.3 | |
| 7 | 3.29 ddd | | 13, 12x, 12y, 11x, 11y, 6 , 9, 5' | 65.5 | |
| 9 | 3.86 s | | 2" , 6" , 7, 6, 12x, 12y, 4 β , 4α , 13 | 78.4 | |
| 10α | 3.63 d | $[J_{10\beta,10\alpha}=15.6]$ | | 45.8 | |
| 10β | 4.32 d | | 10α, 11x, 2", 6" | | |
| 11x | 3.34 dd | $[J_{11x,7}=6.6]$ | 11y, 7, 13, 12y, 10β | 61.8 | |
| 11y | 3.45 dd | $[J_{11y,7}=2.4]$ $[J_{11y,11x}=11.0]$ | 7, 11x, 12y, 13, 5' | | |
| 12x | 2.58 dq | | 13, 12y , 7, 9, 2", 6" | 44.0 | |
| 12y | 2.69 dq | | 13, 12x , 11y, 9 | | |
| 13 | 0.85 t | | | 9.0 | |
| NMe | 3.65 s | | | 29.5 | |
| 1" | | | | 140.0 | 9, 3", 5" |
| 2", 6" | 7.49 br | | | 129.6 | |
| 3", 5" 4" | 7.32 m | | | 127.3 | |
| | 7.28 m | | | 127.0 | |
| 2' | | | | 137.5 | 10α , 10β , 6, NMe |
| 3' | | | | 106.2 | 10α, 10β, 6 |
| 4' | | | | 127.0 | 6', 8', 6 |
| 5' | 7.57 dd | | | 120.8 | |
| 6' | 7.12 td | | | 119.5 | |
| 7' | 7.19 td | | | 120.8 | |
| 8' | 7.30 dd | | | 109.3 | |
| 9' | | | | 137.0 | 5', 7' |

^a Different characters are used as follows. Small: weak enhancement (n.O.e.<2%); normal: medium enhancement (2%<n.O.e.<6%); bold: strong enhancement (n.O.e. >6%), underlined: small negative n.O.e. (three-spin effect). ^b The indicated protons are those having $^{2-3}J_{C,H}$ correlations with the relevant quaternary carbon as detected in an HMBC experiment.

Table 2 ¹H and ¹³C n.m.r. chemical shifts, selected H-H couplings measured n.O.e. and HMBC connectivities for compound 14

Figure 1 Schematic geometrical representation of 11. (For clarity unlabelled bonds denote H).

Compound 14. 1 H and 13 C n.m.r. data for compound 14 are collected in Table 2. Assignments were confirmed as noted above in the case of compound 11. The stereostructure of 14, as derived from the measured H-H n.O.e.s (Table 1), is shown in Fig. 2. From the n.O.e. results the depicted conformation of the CH₂OH moiety appears to be dominant and accords with the relevant $J_{\text{H-11,H-7}}$ vicinal couplings. However, the N-CH₂-CH₃ sidechain has two significantly populated conformations: one as shown in Fig. 2 and the other with C(13) being antiparallel with C(9). As compared to 11 the geminal coupling $J_{10\beta,10\alpha}$ in 14 is larger (15.6 Hz) since the above-noted antiperiplanar effect of the N lone pair is missing.

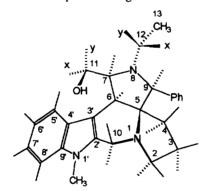


Figure 2 Schematic geometrical representation of 14 (For clarity unlabelled bonds denote H).

EXPERIMENTAL PART

Methods. Column chromatography was performed using Merck Kieselgel 60 70-230 mesh, TLC on aluminium sheets coated with Kieselgel 60 F_{254} . Plates were stained with anisaldehyde solution (100 ml glacial acetic acid, 2 ml cc. sulphuric acid and 1 ml anisaldehyde) and heated at ca. 150° C. IR spectra were measured on a

NICOLET FT-IR instrument. Low resolution electron impact mass spectra were obtained on a Varian CH5-D spectrometer. Mass spectra were run on a VG TRIO-2 AEI-MS-902 (70 eV; direct insertion). N.m.r. measurements were carried out on Varian UNITYplus 500 instrument (500 MHz for 1 H and 125 MHz for 13 C) at 30 $^{\circ}$ C in CDCl₃. Chemical shifts are given relative to δ_{TMS} =0.00 ppm. The DQFCOSY, HSQC, HMBC and NOE experiments were recorded by using the standard spectrometer software package. NOEs were measured in non-degassed samples with 10 s preirradiation times. The presence of all observed NOEs was further verified on a Varian VXR-300 instrument (300 MHz for 1 H) at 24°C. Melting points are uncorrected.

1-Acetyl-2-ethoxycarbonyl-3-(N-methylindol-3-yl)-4-nitro-5-phenyl-pyrrolidine (7). Acetic anhydride 1.2 mL (1.30 g, 14 mM) was added at 0 °C to a solution of 6 adduct⁸ (2.0 g, 5.1 mM) in pyridine (5 mL). The mixture was stirred at room temperature for 2 h, then was poured into ice-water. The products was extracted with CH₂Cl₂ (3x15 mL), then the organic layer was washed sequentially with 3 % HCl (3x10 mL), saturated aqueous NaHCO3, and brine. Then was dried over MgSO4, and concentrated in vacuo. The residue was trituated with ether afforded crystalline product as white powder 2.0 g (92 %); m.p. 183 °C; H-NMR (300 MHz) δ: 7.71 (d, 2H, Ph-2H and Ph-6H), 7.60 (d, 1H, Ind-4H), 7.47-7.37 (m, 3H, Ph-3H, Ph-4H, Ph-5H), 7.28-7.20 (m, 2H, Ind-6'H, Ind-7'H), 7.15 (t, 1H, Ind-5'C), 6.91 (Ind-2'H), 5.69 (dd, 1H, J=8.5 and 12.1 Hz, H-4), 5.40 (d, 1H, J=8.5 Hz, H-5), 5.31 (d, 1H, J=9.0 Hz, H-2), 4.68 (dd, 1H, J=9.8 and 11.5 Hz, H-3), 3.71 (q, 2H, CH₂), 3.66 (s, 3H, NMe), 1.69 (s, 3H, Ac), 0.45 (t, 3H, CH₂); ¹³C-NMR (75 MHz) δ: 170.9 (C=O), 170.7 (C=O), 138.2 (Ind-7a'C), 137.0 (Ph-1'C), 129.5 (Ph-3'C and Ph-5'C), 129.2 (Ph-4'C), 127.3 (Ind-3a'C), 126.8 (Ph-2'C and Ph-6'C), 126.2 (Ind-2'C), 122.4 (Ind-6'C), 119.7 (Ind-5'C), 118.8 (Ind-4'C), 109.3 (Ind-7'C), 105.0 (Ind-3'C), 94.2 (C-4), 66.8 (C-2), 63.1 (C-5), 60.8 (CH₂), 42.2 (C-3), 32.7 (CH₃N), 22.7 (CH₃CO), 12.8 (CH₃CH₂); IR (KBr, cm⁻¹): 3447, 3068, 2963, 1740, 1653, 1549, 1487, 1458, 1395, 1377, 1341, 1262, 1245, 1206, 1182, 1096, 1021; MS m/z (rel. intensity %): 435 (M⁺, 53), 389 (5), 345 (33), 315 (20), 301 (18), 273 (36), 257 (10), 246 (20), 230 (8), 216 (11), 202 (8), 168 (22), 154 (12), 144 (27), 131 (45), 115 (48), 103 (8), 91 (21), 77 (23), 65 (5), 51 (9), 43 (base peak); Anal. calcd. for C₂₄H₂₅N₃O₅: C 66.19, H 5.79, N 9.65; found C 66.12, H 5.65, N 9.63

1-Acetyl-2-ethoxy carbonyl-4-(2-methoxy carbonylethyl)-3-(N-methyl indol-3-yl)-4-nitro-5-phenyl-4-(2-methoxy carbonylethyl)-3-(N-methyl indol-3-yl)-4-nitro-5-phenyl-4-(2-methoxy carbonylethyl)-3-(N-methyl indol-3-yl)-4-nitro-5-phenyl-4-(2-methoxy carbonylethyl)-3-(N-methyl indol-3-yl)-4-nitro-5-phenyl-4-(2-methoxy carbonylethyl)-3-(N-methyl indol-3-yl)-4-nitro-5-phenyl-4-nitr

pyrrolidine (8). Compound 7 (2.0 g 4.6 mM) was dissolved in 30 mL acetonitrile, 2.0 mL (1.9 g, 2.2 mM) methyl acrylate and 0.2 mL Triton B (*N*-benzyltrimethylammonium-hidroxide) were added. The mixture was stirred at room temperature under nitrogen for 48 h, and then poured into dilute HCl (15 mL), and CHCl₃ (35 mL). The aqueous layer was separated and further extracted with CHCl₃ (3x25 mL). The combined organic exracts were washed sequentially with 15 mL portions of saturated NaHCO₃ and brine, dried over MgSO₄,

filtered, and evaporated *in vacuo* to yield a white powder 1.63 g (68 %); m.p.193 °C; ¹H-NMR (300 MHz) 8: 7.58 (d, 1H, Ind-4'H, 7.40 (bs, 2H Ph-2'H and Ph-6'H), 7.36-7.29 (m, 3H, Ph-3'H, Ph-4'H, Ph-5'H), 7.28-7.20 (m, 2H, Ind-6'H and Ind-7'H), 7.19 (t, 1H, Ind-5'H), 6.63 (s, 1H, Ind-2'H), 5.29 (d, 1H, *J*= 10.0 Hz, H-2), 5.20 (s, 1H, H-5), 4.29 (d, 1H, *J*=10.0 Hz, H-3), 4.26-4.04 (m, 2H, CH₂CH₃), 3.74 (s, 3H, NMe), 3.32 (s, 3H, CO₂Me), 2.79 (m, 1H) and 2.56 (m, 1H), and 2.18 (m, 2H, CH₂CH₂), 1.75 (s, 3H, Ac), 1.00 (t, 3H, CH₃CH₂); ¹H-n.O.e. (%): irradiation of H-3 caused enhancement of Ind-4'H (11.8) and H-2 + H-5 (13.0) and CH₂CH₂CO₂Me (3.6 and 4.4); ¹³C-NMR (75 MHz) 8: 172.2 (C=O), 170.7 (C=O), 168.7 (C=O), 136.0 (Ind-7a'C), 135.4 (Ph-1'C), 129.4 and 129.1 (Ph-2'C and Ph-6'C), 128.8 (Ph-4'C), 128.5 (Ind-2'C), 127.4 and 126.5 (Ph-3'C and Ph-5'C), 122.2 (Ind-6'C), 119.8 (Ind-5'C), 117.5 (Ind-4'C), 109.5 (Ind-7'C), 104.0 (Ind-3'C), 101.0 (C-4), 72.4 (C-5), 62.7 (C-2), 60.7 (CH₂CH₃), 51.7 (OCH₃), 45.0 (C-3), 33.0 (NMe), 31.1 (CH₂), 29.2 (CH₂), 22.9 (CH₃CO), 13.7 (CH₂CH₃); IR (KBr, cm⁻¹): 3446, 2960, 2926, 1735, 1654, 1552, 1438, 1390, 1378, 1336, 1260, 1204, 1179, 1095, 1018; MS *m/z* (rel. intensity %): 521 (M⁺, 78), 490 (4), 433 (5), 429 (13), 401 (18), 387 (8), 359 (20), 327 (8), 301 (9), 285 (24), 272 (13), 244 (14), 228 (15), 214 (12), 182 (14), 168 (24), 157 (23), 146 (base peak), 131 (58), 117 (30), 105 (10), 91 (23), 77 (15), 55 (16), 43 (98); Anal. calcd. for C₂₈H₃₁N₃O₇: C 64.48, H 5.99, N 8.06; found C 64.32, H 5.85, N 8.00.

8-Acetyl-7-ethoxycarbonyl-6-(N-methylindol-3-yl)-9-phenyl-2-oxo-1,8-diazaspiro-[4,4]-nonane (9). To a stirred solution of nitro-ester 8 (1.50 g, 2.9 mM) in ethanol (150 mL) was addded zinc dust (3.3 g, 50 mM). This mixture was heated to 40-45 Co then concentrated HCl (4 mL) was dropped meanwhile the temperature was kept between 45-50 °C. The reaction mixture was then refluxed for 12 h, filtered, evaporated in vacuo nearly to dryness when the residue was dissolved in saturated NaHCO₃ solution (20 mL) and CH₂Cl₂ (40 mL), filtered and extracted in CH₂Cl₂ (3x50 mL). The combined organic extracts was washed with brine, dried, and evaporated in vacuo to yield a white solid 1.05 g (80%); m.p. 236 °C; H-NMR (300 MHz) δ: 7.62 (d. 1H. Ind-4'H), 7.41-7.25 (m, 7H, Ph and Ind-6'H, Ind-7'H), 7.19 (t, 1H, Ind-5'H), 6.94 (s, 1H, Ind-2'H), 5.02 (s, 1H, H-9), 5.01 (d, 1H, J= 10.0 Hz, H-7), 4.09 (d, 1H, J=10.0 Hz, H-6), 4.03 (q, 2H, CH, CH,), 3.75 (s, 3H, NMe), 2.24 (m, 1H, H-3), 2.12 (m, 1H, H-3), 1.67 (m, 2H, H-4), 1.64 (s, 3H, Ac), 0.88 (t, 3H, CH,CH,); ¹³C-NMR (75 MHz) 8: 136.7 (Ind-7a'C), 136.1 (Ph-1'C), 128.9 (Ph-2'C and Ph-6'C), 128.6 (Ph-4'C), 128.3 (Ind-2'C), 127.7 (Ind-3a'C), 127.4 (Ph-3'C and Ph-5'C), 122.3 (Ind-6'C), 119.7 (Ind-5'C), 117.7 (Ind-4'C), 109.4 (Ind-7'C), 104.4 (Ind-3'C), 73.4 (C-4), 72.1 (C-5), 64.4 (C-2), 61.3 (CH₂CH₃), 45.5 (C-3), 33.0 (NMe), 29.1 (CH₂), 28.7 (CH₂), 23.5 (CH₃CO), 13.6 (CH₃CH₂); IR (KBr, cm⁻¹): 3373, 3128, 3057, 2978, 2934, 1699, 1651, 1544, 1474, 1456, 1398, 1381, 1333, 1309, 1280, 1238, 1209, 1108, 1066, 1026, 1014; MS m/z (rel. intensity %): 459 (M⁺,22), 413 (10), 406 (17), 375 (12), 343 (16), 311 (31), 266 (72), 229 (85), 226 (52), 197 (48), 169 (39), 157 (76), 144 (77), 131 (22), 117 (80), 91 (20), 77 (35), 65 (21), 55 (29), 43 (base peak); Anal. calcd. for C₂₇H₂₉N₃O₄: C 70.57, H 6.36, N 9.14; found C 70.61, H 6.28, N 9.11.

8-Ethyl-7-(hydroxymethyl)-6-(N-methylindol-3-vl)-9-phenyl-1,8-diazaspiro-[4.4]-nonane (10). Lactam 9 (0.8 g, 1.7 mM) dissolved in 8 mL THF was added to a stirred suspension of LiAlH₄ (1.0 g, 26 mM) in rigoruosly dried THF (15 mL, distilled from potassium) under nitrogen. The reaction mixture was then refluxed for 24 h, cooled, and treated with water (2 mL) and 20% NaOH solution (4 mL). The mixture was warmed to room temperature, filtered and the filtrate was washed sequentially with 20% NaOH solution (15 mL), brine, then dried over MgSO₄, and concentrated to yield compound as white foam 0.55 g (72 %); ¹H-NMR (250 MHz) δ: 7.66 (d, 1H, Ind-4'H), 7.6 (bs, 2H, Ph-2'H, Ph-6'H), 7.40 (m, 3H, Ph-3'H, Ph-4'H, Ph-5'H), 7.34 (m, 2H, Ind-7'H, Ind-2'H), 7.24 (m, 1H, Ind-H'6), 7.14 (m, 1H, Ind-H'5), 4.26 (s, 1H, H-9), 4.05 (d, 1H, J=9.8 Hz, H-6), 3.80 (s, 3H, NMe), 3.73 (dd, 1H, J=2.8 Hz, and 11.3 Hz, CH,-OH), 3.61 (dd, 1H, J=1.3 and 11.3 Hz, CH,-OH), 3.26 (ddd, 1H, J=1.3 and 2.8 and 9.8 Hz, H-7), 2.79 (m, 1H, H-2), 2.67 (m, 1H, H-2), 2.31 (q, 2H, CH,-CH₂), 1.84 (m, 1H, H-4), 1.70 (m, 1H, H-4), 1.05 (t, 3H, CH₂), 0.95 (m, 1H, H-3), 0.80 (m, 1H, H-3), ¹H-n.O.e. (%): irradiation of H-9 caused enhancement of H-6 (7.3) and H-2 (0.8 + 1.5) and H-4 (3.9) and H-3 (2.4) and Ph-2'6'H (8.7); irradiation of H-6 caused enhancement of Ind-4'H (13.9) and H-9 (8.8) and H-7 (3.0) and H-4 (3.2); irradiation of H-7 caused enhancement of Ph-2'6'H (2.1) and H-6 (7.2), CH₂OH (1.3 and 1.4) and H-2 (0.7); (proof of stereochemistry; medium large n.O.e. enhancement on H-4 at the irradiation of H-9 or H-7). ¹³C-NMR (75 MHz) δ: 139.0 (Ind-7a'C), 136.1 (Ph-1'C), 129.9 Ind-3a'C), 129.4 (Ind-2'C), 128.5 (Ph-2'C and Ph-6'C), 128.3 (Ph-4'C), 127.8 (Ph-3'C and Ph-5'C), 121.2 (Ind-6'C), 118.9 (Ind-5'C), 118.0 (Ind-4'C), 109.1 (Ind-7'C), 108.0 (Ind-3'C), 75.1 (C-9), 72.1 (C-5), 64.4 (C-7), 63.1 (CH₂OH), 46.6 (C-2), 32.9 (NCH₃), 32.5 (C-4), 26.2 (C-3), 11.7 (CH₃); IR (KBr, cm⁻¹): 3281, 3112, 3050, 2959, 2931, 2873, 1716, 1681, 1612, 1582, 1555, 1471, 1460, 1422, 1375, 1331, 1265, 1234, 1157, 1132, 1071, 1042, 1012; MS m/z (rel. intensity %): 389 (M⁺,10), 330 (4), 284 (7), 255 (10), 227 912), 212 (40), 197 (12), 176 (35), 158 (49), 144 (50), 131 (37), 118 (37), 104 (18), 91 (base peak), 77 (22), 56 (15), 41 (52); Anal. calcd. for C₂₅H₃₁N₃O: C 77.08, H 8.02, N 10.79; found C 77.17, H 8.08, N 10.74.

Compound 11. Amine 10 (0.40 g, 1 mM) and paraformaldehyde (0.40 g 13.6 mM) was dissolved in benzene (25 mL), and one drop of concentrated HCl was added. After 3 h reflux the cooled mixture was poured into saturated NaHCO₃ solution, extracted into CHCl₃, the extract was dried over MgSO₄ and evaporated to yield 0.32 g (79.8 %) white foam; ¹H-NMR and ¹³C-NMR see Table 1; IR (KBr, cm⁻¹): 3024, 2962, 2933, 2867, 2792, 1606, 1491, 1466, 1452, 1388, 1336, 1296, 1271, 1255, 1210, 1160, 1105, 1082, 1021, 981, 946, 757, 703; MS m/z (rel. intensity %): 401 (M⁺, 18), 371 (13), 267 (15), 230 (52), 223 (80), 198 (25), 182 (10), 169 (14), 157 (13), 144 (base peak), 104 (16), 91 (41), 77 (24), 55 (32), 43 (26). Anal. calcd. for C₂₆H₃₁N₃O: C 77.77, H 7.78, N 10.46; found C 77.83, H 7.86, N 10.38.

7-(tert-Butyldimethylsilyloxymethyl)-8-ethyl-6-(N-methylindol-3-yl)-9-phenyl-1,8-diaza-

spiro[4.4]nonane (12). Amine 10 (1.00 g 2.6 mM) and TBDMSCl (0.48 g, 2.8 mM) was dissolved in 8 mL acetonitrile and DBU (0.59 g, 3.9 mM) was added. The reaction mixture was stirred at room temperature for 24 h then 3 mL water and 15 mL ether were added. The organic phase was washed with brine, dried, and evaporated *in vacuo* to yield an oil which was purified by flash chromatography (eluent: hexane- ethyl acetate 1:1) to give a colorless oil 1.15 g (88 %); ¹H-NMR (500 MHz, CDCl₃) δ: 7.64 (d, 1H, Ind-4H), 7.55 (bs, 3H, Ph-2', 4', 6'H), 7.35 (t, 2H, Ph-3' and 5'H), 7.31 (m, 1H, Ind-7'H), 7.27 (s, 1H, Ind-2'H), 7.20 (t, 1H, Ind-5'H), 7.11 (t, 1H, Ind-6'H), 4.03 (s, 1H, H-9), 3.81 (d, 1H, J = 9 Hz), H-6), 3.77 (s, 3H, NMe), 3.62 (m, 2H, CH₂OH), 2.27 (td, 1H, H-7), 2.90 (dq, 1H, N-CH₂), 2.67 (dq, 1H, NCH₂), 2.57 (dt, 1H, H-2), 1.87 (m, 2H, H-2, H-4), 1.76 (m, 2H, H-3, H-4), 1.03 (m, 1H, H-3), 0.96 (t, 3H, CH₃), 0.94 (s, 9H, Bu¹), 0.11 (s, 6H, SiMe₂); ¹³C-NMR (125 MHz, CDCl₃) δ: 138.8 (Ind-7a'C), 136.0 (Ph-1'C), 130.3 (Ind-3a'C), 129.5 (Ind-2'C), 128.5 (Ph-2'C and 6'C), 127.9 (Ph-3'C and 5'C), 127.2 (Ph-4'C), 121.3 (Ind-6'C), 118.9 (Ind-5'C), 118.0 (Ind-4'C), 109.2 (Ind-7'C), 108.8 (Ind-3'C), 76.1 (C-9), 66.4 (C-5), 64.4 (C-7), 63.0 (CH₂OH), 48.9 (C-6), 47.6 (C-2), 46.5 (NCH₂), 33.4 (C-4), 33.0 (NCH₃), 32.5 (CMe₃), 25.8 (CMe₃ + C-3), 11.0 (CH₃); IR (film, cm⁻¹): 3382, 3056, 3025, 2955, 2931, 2855, 1613, 1420, 1374, 1329, 1254, 1155, 1092, 836; Anal. calcd. for C₃₁H₄₅N₃OSi: C 73.91, H 9.00, N 8.34; found C 73.81, H 9.08, N 8.34.

1-(tert-Butyldimethylsililoxymethyl)-2-ethyl-8-methyl-3-phenyl-1, 2, 3, 4, 5, 6, 7, 12b-octahydro-density and the state of the sta

pyrrolo[3',4':5,6]pyrrolo[1',2':6,1]pyrimido[3,4-b]indole (13). TBDS-amine 12 (0.30 g, 0.59 mM) and Eschenmoser salt (0.21 g, 1.1 mM) were dissolved in 6 mL dry THF. After 12 h stirring at room temperature the reaction mixture was poured into 5 mL saturated Na₂S₂O₃ solution, and extracted with 30 mL ether. The organic extract was washed with brine, dried, and evaporated *in vacuo* to give an oil which was purified by column chromatography (eluent: hexane- ethyl acetate 1:1) to yield 0.21 g (69%) colorless oil; ¹H-NMR (500 MHz, CDCl₃) δ: 7.86 (d, 1H, H-5'), 7.60 (bs, 2H, H-2" and H-6"); 7.49-7.39 (m, 4H, H-3", 4", 5" and H-8"), 7.31 (td, 1H, H-7'), 7.23 (td, 1H, H-6'), 4.21 (d, 1H, J = 16.1 Hz, H-10), 3.96 (s, 1H, H-9), 3.83 (dd, 1H, J = 5.5 and 10.7 Hz, CH₂OH), 3.73 (s, 1H, NMe), 3.67 (d, 1H, J = 15.8Hz, H-10), 3.70 (d, 1H, J = 8.3 Hz, H-6), 3.35 (td, 1H, J = 8.3 and 5.4 Hz, H-7), 2.98 (dq, 1H, NCH₂), 2.85 (dt, 1H, J = 8.3 and 3.4 Hz, H-2), 2.80 (dq, 1H, NCH₂), 2.34 (m, 2H, H-2 and H-4), 2.04 (td, 1H, J = 7.5 and 12.5 Hz, H-4), 1.45 (m, 1H, H-3), 1.05 (m, 1H, H-3), 0.98 (s, 9H, Bu¹), 0.97 (t, 3H, CH₃), 0.06 (s, 6H, SiMe₂); ¹³C-NMR (125 MHz, CDCl₃) δ: 140.3 (C-1"), 136.8 (C-2' and C-9'), 129.6 (C-2" and C-6"), 128.4 (C-4'), 127.1 (C-3" and C-5"), 126.6 (C-4"), 120.2 (C-7'), 118.7 (C-6'), 118.6 (C-5'), 108.5 (C-8'), 107.6 (C-3'), 77.0 (C-9), 71.4 (C-5), 66.8 (C-7), 64.2 (CH₂OH), 54.7 (C-2), 44.1 (C-10), 43.6 (NCH₂), 42.8 (C-6), 40.6

(C-4), 29.7 (CMe₃), 29.3 (NMe), 25.9 (CH₃CH₂), 22.4 (C-3), 8.3 (CMe₃); IR (film, cm⁻¹): 2960, 2919, 2954, 1469, 1308, 1258, 1167, 1096, 1048; Anal. calcd. for C₃₂H₄₅N₃OSi: C 74.51, H 8.79, N 8.15; found C 74.39, H 8.78, N 8.13.

2-Ethyl-1-(hydroxy-methyl)-8-methyl-3-phenyl-1,2,3,4,5,6,7,12b-octahidro-pyrrolo[3',4':5,6]

pyrrolo[1',2':6,1]pyrimido[3,4-b]indole (14). 13 (0.1 g, 0.19 mM) was dissolved in 5 mL THF and Bu₄NF on silica (0.5 g, 1.1 mM/g, 0.55 mM) was added. The reaction mixture was stirred at room temperature for 48 h, then the catalyst was removed by filtration. The solvent was evaporated *in vacuo* to yield an oil which was purified by column chromatography furnishing a colorless oil crystallized very slowly on standing to give a white powder; 72 mg (95 %). 1 H-NMR and 13 C-NMR see Table 2; IR (KBr, cm $^{-1}$): 3416, 2942, 2926, 1466, 1373, 1315, 1240, 1216, 1200, 1130, 1971, 1044; MS m/z (rel. intensity %): 401 (M $^{+}$, 12), 370 (14), 313 (14), 233 (base peak), 208 (6), 176 (13), 146 (10), 91 (12), 55 (5); Anal. calcd. for $C_{26}H_{31}N_3O$: C 77.77, H 7.78, N 10.46; found C 77.79, H 7.74, N 10.43.

Acknowledgements

This work was financially supported by the *National Fund for Science and Research* (OTKA Project No. 2461). A grant from the *Jözsef Varga Foundation* provided to M.Ny. is gratefully appreciated.

REFERENCES

- Huang, L. Xue, Z. 'The Alkaloids. Chemistry and Pharmacology'; Vol.23. ed. by Brossi, A. Academic Press: New York. 1984; pp. 157-226.
- (a) Auerbach, J.; Weinr1eb, S. M. J. Am. Chem. Soc. 1972, 94, 7172. (b). Auerbach, J.; Weinreb, S. M. J. Am. Chem. Soc. 1975, 97, 2503. (c) Semmelhack, M.F.; Chong, B. P.; Jones, L. D. J. Am. Chem. Soc. 1972, 94, 8629. (d) Semmelhack, M. F.; Stauffer, R. D.; Rogerson, T. D. Tetrahedron Lett. 1973, 14, 4519. (e) Semmelhack, M. F.; Chong, B. P.; Stauffer, R.D.; Rogerson, T. D.; Chong, A.; Jones, L. D. J. Am. Chem. Soc. 1975, 97, 2507. (f) Yasuda, S.; Yamada, T.; Hanaoka, M. Tetrahedron Lett. 1986, 27, 2023. (g) Kuehne, M.E.; Bornmann, W. G.; Parsons, W. H.; Spitzer, T. D.; Blount, J. F.; Zubieta, J. J. Org. Chem. 1988, 53, 3439. (h) Burkholder, T.P.; Fuchs, P. L. J. Am. Chem. Soc. 1988, 110, 2341. (i) Ikeda, M.; Kosaka, K.; Sakakibara, M.; Okamo, M. Heterocycles 1993, 35, 81. (j) Ishibashi, M.; Okano, M.; Tanaki, H.; Marayuma, K.; Yakura, T.; Ikeda, M. J. Chem. Soc. Chem. Comm. 1990, 1436. (k) Ikeda, M.; Okano, M.; Kosaka, K.; Kido, M.; Ishibashi, M. Chem. Pharm. Bull. 1993, 41, 276. (l) Lin, X.; Kavash, R.W.; Mariano, P.S. J. Am. Chem. Soc. 1994, 117, 9791. (m) Bailey, P.D.; Morgan, K.M.; Smith, D.I.; Vernon, J.M. Tetrahedron Lett. 1994, 35, 7115. (n) Ribeiro

- C.M.R.; Melo, S.J.; Bonin, M.; Quirion J.-C.; Husson, H.P.; *Tetrahedron Lett.* 1994, 35, 7227. (o) Schinzer, D.; Langkopf, E. *SYNLETT* 1994, 375. (p) Ikeda, M.; Matsubayashi, K.; Imoto, T.; Kitao, K.; Ishibashi, H.; Sato, T.; *Heterocycles* 1994, 38, 1237. (q) Isomo, M.; Mori, M. *J. Org. Chem.* 1995, 60, 115.
- 3. Zhou, J. Y.; Chen, D. L.; Shen, Z.; Koeffer, H. P. Cancer Res. 1990, 50, 2031.
- 4. Gauvin-Hussenet, C.; Seraphin, B.; Cartier, D.; Laronze, J.X.; Levy, J. Tetrahedron Lett. 1993, 34, 465.
- (a) Hill, R. K.; Sawada, S.; Rock, M. G.; Greene, J. R. Heterocycles 1987, 25, 515. (b) Gardiner, J. M.;
 Bryce, M. R. Tetrahedron 1988, 44, 599. (c) Gardiner, J. M.; Bryce, M. R. J. Chem. Soc. Chem.
 Comm. 1989, 1162. (d) Gardiner, J. M.; Bryce, M. R. J. Org. Chem. 1990, 55, 1261.
- Okano, M.; Nishimura, N.; Maruyama, K.; Kosaka, K.; Ishibashi, M.; Ikeda, M. Chem. Pharm. Bull. 1991, 39, 3163
- (a) Nyerges, M.; Bitter, I.; Kádas, I.; Tóth, G.; Tőke, L. Tetrahedron Lett. 1994, 35, 4413. (b) Nyerges,
 M.; Bitter, I.; Kádas, I.; Tóth, G.; Tőke, L. Tetrahedron 1995, 51, 11489.
- 8. Nyerges, M.; Rudas, M.; Tóth, G.; Herényi, B.; Kádas, I.; Bitter, I.; Tőke, L. *Tetrahedron* 1995, 51, 13321.
- 9. Tsuge, O.; Kanemasa, S. 'Adv. in Heterocyclic Chemistry' Vol.45. ed. by Katritzky, A.; Academic Press 1989, pp. 232-349.
- 10. Schreiber, J.; Maag, H.; Hashimoto, N.; Eschenmoser, A. Angew. Chem. Int. Ed. Engl. 1971, 10, 330.
- (a) Jackson, R.H.; Smith, P. Tetrahedron 1968, 24, 2227. (b) Ungemach, F.; Cook, J.M.; Heterocycles 1978, 9, 1089. (c) Bailey, P.D.; Hollinshead, S.P.; McLay, N.R.; Morgan, K.; Palmer, S.J.; Prince, S.N.; Reynolds, C.D.; Wood, S.D. J. Chem. Soc. Perkin Trans. 1. 1993, 431. (d) Casnati, G; Dossena, A.; Pochini, A. Tetrahedron Lett. 1972, 13, 5277. (e) Cox, E.D.; Cook, J.M. Chem. Rev. 1995, 95, 1797.
- 12. Chivers, P. J.; Crabb, T. A. Tetrahedron 1970, 26, 3389.

(Received in UK 26 November 1996; revised 31 December 1996; accepted 9 January 1997)